Biology Knowledge Organiser B9 - Respiration

Respiration.

Photosynthesis produces chemicals (like glucose) that can be used as food by all living organisms. In **respiration**, the chemical potential energy stored in food molecules is transferred through **oxidation** reactions (where oxygen, originally from the air, reacts with the food molecules). The energy transferred allows living cells to do **work**.

As you know, doing work means transferring energy. The kinds of work done by cells and organisms include:

- Chemical reactions to build larger molecules from smaller ones. E.g. making proteins such as enzymes from amino acids.
- Movement. E.g. movements of our body are possible due to muscle contractions. This requires energy from respiration.
- Keeping warm. This is an example of homeostasis: using energy from respiration to maintain body temperature at a set point (37°C).

There are two types of respiration: aerobic and anaerobic.

Aerobic vs. Anaerobic Respiration.

Aerobic respiration occurs when oxygen is used in the reaction. It is shown by these equations:

glucose + oxygen \rightarrow carbon dioxide + water C₆H₁₂O₆ + 6O₂ \rightarrow 6CO₂ + 6H₂O

This reaction releases energy that can be used by organisms, as described above. Compared to anaerobic respiration, aerobic respiration releases much more energy.

<u>Anaerobic respiration</u> occurs when there is insufficient oxygen available for complete oxidation of the glucose. The reaction that happens is different in animal cells compared to plant and yeast cells.

In *animals*: In *plants* and *yeast*: glucose → lactic acid glucose → ethanol and carbon dioxide $C_6H_{12}O_6 \rightarrow 2C_2H_5OH + 2CO_2$

Anaerobic respiration releases much less energy than aerobic respiration. In yeast, we call the anaerobic respiration **fermentation**. This is a very useful process: for making bread (the CO_2 makes it rise) and making alcoholic drinks (since ethanol is a type of alcohol).

Key Terms	Definitions
Aerobic	Using oxygen
Anaerobic	Not using oxygen
Oxidation	A reaction with oxygen. In this case, food molecules like glucose reacting with oxygen.
Fatigue	Tiredness. Fatigue in muscles is caused by a build-up of lactic acid, which is produced during anaerobic respiration (when there is insufficient oxygen).
Oxygen debt	After exercise, the lactic acid has built up and caused an extra need for oxygen – called the oxygen debt.
Lactic acid	Chemical produced by the incomplete oxidation of glucose (anaerobic respiration).

The Response To Exercise.

During exercise, more energy is required by the body that when resting, due to increased muscle contractions. The body reacts to this increased **demand** for energy:

The <u>heart rate</u>, <u>breathing rate</u>, and <u>volume of each breath</u> all increase. Together, these increase the amount of **oxygenated blood** reaching the muscles. The oxygenated blood provides the extra oxygen and glucose needed for respiration in muscle cells, to transfer more energy to meet demand.

However, if insufficient oxygen reaches muscles but exercise continues, the muscle cells use **anaerobic respiration** to transfer energy. From the equation, you can see that incomplete oxidation of glucose takes place and **lactic acid** is produced. The lactic acid builds up and causes an **oxygen debt**. The lactic acid building up also causes **fatigue**. Removing the lactic acid after exercise is the cause of the oxygen debt – the oxygen debt is why you breathe deeply after exercise for some time. You are 'repaying' the oxygen debt.

HT: oxygen debt, to be precise, is the amount of extra oxygen needed to react with lactic acid in muscles and remove it from cells. The blood flow through muscles removes lactic acid and transports it to the liver. In the liver, the lactic acid is converted back into glucose. This reaction requires energy, hence the extra need for oxygen (for aerobic respiration to provide that energy).

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Metabolism.

Metabolism is a very big term in biology. It is the name given to collectively describe ALL the <u>chemical reactions</u> happening in a cell or in the whole body. So, respiration in all cells is an example of metabolism, and so is photosynthesis in plants.

Many reactions that cells perform require **energy**, so metabolism relies on energy transferred by respiration. Furthermore, chemical reactions in cells are controlled by **enzymes**. As we're talking about chemical reactions, *reactants* are used to make *products*: new molecules are synthesised.

3 Fatty Acids + Glycerol

Enzyme type :

Enzyme type 2

of enzyme 2

To learn: metabolism includes these reactions:

- Conversion of glucose to glycogen (in animals), or tostarch or cellulose (in plants).
- Making lipid (fat) molecules from one molecule of **glycerol** and three molecules of **fatty acids** (see diagram).
- In plants, the use of glucose and nitrate ions to make amino acids. These amino acids are then used to synthesise proteins.
- Respiration, both aerobic and anaerobic.
- Breaking down excess proteins into amino acids, then into **urea** for excretion in the urine.

Enzyme type 1

Factors Affecting Enzymes.

Recap your knowledge of how enzymes work from Topic 7.

Enzymes are highly **specific**, meaning that each type of enzyme only causes a reaction by one type molecule.

This comes about due to the specific shape of the active site: only one molecule (according to its shape) will fit into the active site. See diagram for an illustration.

Enzymes have an optimum temperature and pH. If the temperature is **too high** (for most enzymes, above about 45°C), or the pH is **too acidic** OR **too alkaline**, the enzyme **denatures**. This means the active site changes shape. As a result, the substrate no longer fits into the active site, to the enzyme doesn't work any more (see diagram).

This leads to results with enzyme controlled reactions as shown in the graphs. The rate of the reaction catalysed by the enzyme is on the y-axis. The peak represents the optimum temperature/pH. Notice that different enzymes have different optimums – as shown on the pH graph with the two lines.

1	Key Terms	Definitions
	Metabolism	The sum of all the chemical reactions in a cell or in the body of an organism.
	Enzyme	Large protein molecule that acts as a biological catalyst, dramatically speeding up chemical reactions in organisms.
	Synthesis	Making something new. E.g. new molecules in metabolism.
	Active site	The part of an enzyme molecule into which the substrate fits – so the shape of the active site is vital.
	Substrate	The molecule an enzyme 'works on' to make a product/products.
4	Optimum	The ideal or perfect condition . Enzymes have an optimum temperature and an optimum pH.
	Active site I 2 PH	Temperature too high or pH too alkaline/too acidic, so enzyme denatured.